

Production, Application, and Efficacy of Biodefoamers from *Bacillus*, *Aeromonas*, *Klebsiella*, *Comamonas* spp. Consortium for the Defoamation of Poultry Slaughterhouse Wastewater

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Abstract: Activated sludge (AS) treatment systems' major limitation is the nuisance foaming at the surface of the aeration basin in wastewater treatment plants (WWTPs). This foam can be stabilized by biofoamers and surfactants in the wastewater to be treated. In order to control foam, synthetic defoamers are used; however, these defoamers are toxic to the environment. This study aimed to optimize the production of biodefoamers by quantifying foam reduction efficiency and foam collapse by the isolate pervasive to poultry slaughterhouse wastewater (PSW). Before their identification and characterization, nine bacterial isolates were isolated and assessed for foam reduction efficiency. These organisms produced minute biodefoamers under various conditions generated on the response surface methodology (RSM). The isolates that produced biodefoamers with high foam reduction efficiency and at a lower foam collapse rate were *Bacillus*, *Aeromonas*, *Klebsiella*, and *Comamonas* groups, which indicated a polysaccharide core structure. The ¹H NMR analysis further confirmed that the biodefoamers were carbohydrate polymers. This study reports for the first time on the efficiency and comparability of a biodefoamer to a synthetic defoamer.

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1. Introduction

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The activated sludge process (ASP) microbiome is a significant constituent that influences the performance of biological wastewater treatment plants (WWTPs). ASP converts or absorbs organics, ammonium nitrogen, and some phosphorous, including suspended or dissolved solids, among many other contaminants [1]. The efficiency of the AS system depends on organisms' ability to form active conglomerates in an aerated environment. This process is influenced by many factors, such as environmental conditions (pH, temperature, dissolved oxygen (DO)) and climate change. Climate change results in increased rainfall and this increases the influent flow rate by 15–25% as compared to dry weather. It also

spp. consortia. At 4% (v defoamer/v PSW), the crude defoamers produced by the consortium had 96% foam reduction efficiency at 1.7 mm/s foam collapse rate, which was comparable to 96% foam reduction efficiency and 2.5 mm/s foam collapse rate for active silicone polymer antifoam A/defoamer by Sigma-Aldrich, a synthetic defoamer. At 2.5 mm/s, all of which were achieved at pH 7 and in less than 50 s. The application of the biodefoamer resulted in sludge compacted flocs, with filament protruding flocs observed when a synthetic defoamer was used. The biodefoamer showed the presence of alkane, amine, carboxyl and hydroxyl

reduces total suspended solids (TSS), chemical oxygen demand (COD) and ammonium nitrogen (N-NH_4^+) concentration due to its dilution effect in the sewer and WWTP.

Although

these systems are designed to carry over capacity during increased precipitation, sludge washout can be encountered, leading to the inefficiency of the system in the long term as well as high energy consumption and will require increased operational costs [2–4]. To curb this problem, storm overflow and combined sewer overflow tanks are used. However, during wet weather flow, the combined sewer may overflow into receiving water bodies and contaminate it; hence, these are short-term solutions. Therefore, an implementable research for long-term solutions is required [3].

When treating wastewater that contains high concentrations of proteins, fat oil and grease (FOG), the system efficiency becomes a challenge because proteins are surface active and they reduce foam drainage, whereas FOG solidifies at lower temperatures, which results in the clogging of pipes and membrane fouling. Furthermore, it enhances filamentous bacteria growth which results in foam formation [5], herein referred to as biofoam. The generation of biofoam at the surface of an AS aeration tank is a nuisance and is thus considered undesirable. Mycolic acid-containing filamentous bacteria are significant contributors to this biofoam [6]. Furthermore, surfactants present in wastewater from various sources, including biosurfactants produced by numerous microorganisms in the AS in combination with aeration, culminate in foam formation. In most instances, biofoam producers result in foam stabilization within WWTPs [7]. The excessive growth of these organisms leads to the production of biosurfactants and sludge deflocculants, which in turn lead to poor sludge settleability and excessive foaming. This increases WWTPs operational costs associated with defoaming operations, culminating in the deterioration of the effluent quality, requiring further treatment and the extension of sludge retention time, which is energy intensive [8]. Periodically, this results in the loss of essential microbiological cells in the AS. This necessitates efficient biological methods that are less energy intensive, which will shorten the sludge retention time and aeration time.

Numerous strategies have been employed to reduce biofoam formation. These strategies include those termed non-specific and specific strategies. Non-specific strategies include using chemical (synthetic) defoamers and physical techniques (water sprays and the adjustment of operational conditions) to remove or reduce foam. In contrast, specific strategies include the use of biological techniques that target the cause of biofoam formation [9].

Using chemical defoamers such as polyaluminum chloride results in sludge disintegration, which inhibits nitrification and increases soluble chemical oxygen demand (COD) concentration, while oil-based defoamers result in overgrowth of the biofoam-producing filamentous bacteria [10]. Furthermore, using chemical defoamers results in their bioaccumulation downstream, whereas applying physical methods requires adjustment or the design of new bioreactors. This necessitates the need to develop new environmentally benign, energy efficient and economic strategies that will use the water constituents to be treated to reduce the excessive growth of biofoamers rather than treating the symptoms, i.e., biofoam.

The literature reviewed reveals that several biological methods have been used successfully to reduce foam-forming filamentous bacteria such that foam formation is minute in AS systems. These methods include the use of biological reactors, such as up-flow packed bed bioreactors, to enhance nitrogen removal and biofilm granulation, as well as aerobic sludge granulation that produces granular sludge that results in nutrient removal and enhanced settling properties and reduces biofoam [9,11]. Bacteriophages that can be produced from various organisms isolated from various wastewaters have also been used, and these bacteriophages were previously determined to improve sludge settleability [10]. A study by Pajdak-Stós et al., [12] elsewhere highlighted that *Lecane inermis* rotifers were observed to “ingest” and decrease the number of branched filamentous bacteria in AS.

This study focused on producing a biodefoamer from poultry slaughterhouse (PSW) consortia isolates for use in PSW biofoam reduction compared to a commonly used synthetic

defoamer. This was assessed using foam reduction efficiency and foam collapse rate. Furthermore, AS floc integrity and maintenance were also assessed microscopically to ascertain agglomeration, and thus settleability, when the produced biodefoamers were used.

2. Materials and Methods

2.1. Isolation and Identification

The isolates were collected using sterile swabs from a PSW discharge spout of a commercial poultry product producer (Cape Town, South Africa). Serial dilutions were performed, and the nutrient agar plates were inoculated using the swabs by a spread plating technique after incubation at 37 °C overnight. After that, they were cultivated into fresh nutrient agar plates, and the agar plates with pure colonies were stored at 4 °C and recultured daily for further experiments. The pure cultures ($n = 9$) were subjected to mixed liquor suspended solids (MLSS) before biofoam generation to determine their efficiency concerning foam reduction efficiency and foam collapse. Some ($n = 4$) of the identified isolates were highly efficient; they were gram stained and viewed under a microscope to determine their gram reaction. The pure culture plates were sent to Inqaba Biotechnical Industries (Pty) Ltd. (Muckleneuk, Pretoria, South Africa) for DNA analysis and identification. The DNA of the pure isolates was extracted using the Quick-DNA Fungal/Bacterial Miniprep Kit (Zymo Research, Irvine, CA, USA, Catalogue No. D6005). The 16S rRNA target region was amplified using OneTaq[®] Quick-Load[®] 2X Master-Mix (New England Biolabs, Ipswich, MA, USA, NEB Catalogue No. M0486) with primers 16S-27F and 16S-1492R with the sequence (5'-3') AGAGTTTGATCMTGGCTCAT and CGGTACCTTGTACGACTT, respectively.

The extracted fragments were sequenced in both the forward and reverse directions using NimaGen BrilliantDye[™] Terminator Cycle Sequencing kit v3.1 (NimaGen B.V., CG Nijmegen, The Netherlands, BRD3-100/1000) and were purified by Zymo Research ZR-96 DNA Sequencing Clean-up Kit[™] (Zymo Research, Irvine, CA, USA, Catalogue No. D4050). The purified DNA fragments were analyzed for each sample on the Applied Biosystems[™] 3500xL Genetic Analyzer (Applied Biosystems, Waltham, MA, USA). A Qiagen CLC Main

Workbench (Hilden, Germany, v7.6) was used to analyze the ab1 files generated by the ABI 3500xL genetic analyzer, and the results were obtained using a Basic Local Alignment Search Tool (BLAST) search provided by the National Center for Biotechnology Information (NCBI, <http://www.ncbi.nlm.nih.gov>, accessed on 4 December 2022).

2.2. Metagenomics Analysis of the MLSS Microbial Community

From the MLSS samples collected using 4 L polypropylene bottles ($n = 9$) from an aeration tank of a WWTP near Cape Town, South Africa, 100 mL of the MLSS samples was filtered through a 0.22 µm millipore membrane filter (Merck Millipore, Burlington, MA, USA) and the cells were resuspended in 5 mL sterile millipore water for DNA extraction. DNA was extracted from the suspension solution using commercially available DNA extraction kits (Promega, Madison, WI, USA) following the manufacturer's instructions. The 16S rRNA target regions were sequenced and amplified using a OneTaq[®] Quick-Load[®]

2X Master-Mix (New England Biolabs, Ipswich, MA, USA, Catalogue No. M0486) with the primers 16S-27 F and 16S-518 R with the sequence (5-3) AGAGTTTGATCMTGGCTCAG and ATTACCGCGGCTGCTGG, respectively. Moreover, the V1 and V3 targeted sequences were used for PCR amplification of the purified DNA sequence. The PCR amplicons were sent for sequencing at Inqaba Biotechnical Industries (Pty) Ltd. (Muckleneuk, Pretoria, South Africa).

The purified PCR amplicons were gel purified and repaired, and the amplicons were sequenced on the PacBio Sequel II system (Pacific Biosciences Inc., Menlo Park, CA, USA). Raw subreads were processed through the SMRT(R) Link (v9.0) Circular Consensus Sequences (CCS) algorithm, which estimates/computes consensus sequences to

produce highly accurate reads (>QV40). These reads were processed through Vsearch (<https://github.com/torognes/vsearch>, accessed on 4 December 2022) (GitHub Inc., San

2TM. The results were acquired through BLAST provided by the National Center for Biotechnology Information (NCBI, <http://www.ncbi.nlm.nih.gov>, accessed on 4 December 2022). The

sequences were further deposited into the national NCBI, and the sequence read archive (SRA) databases before acquiring the accession numbers of single species.

2.3. Biodefoamer Production and Extraction

The pure culture isolates ($n = 4$) growing on nutrient agar plates were all inoculated as a consortium in nutrient broth (50 mL), where conical flasks (250 mL) were used and subsequently incubated at 37 °C for 24 h at 120 rpm (Labwit ZWYR-240 shaking incubator, Labwit Scientific, Burwood East, Vic., Australia) to produce a seed culture. A volume (1 mL) of the 24 h old seed culture in the form of a consortium was inoculated into 99 mL of nutrient broth to a final volume of 100 mL in 250 mL conical flasks and further incubated at a temperature of 37 °C at 120 rpm for 24 h. After that, 15 mL of each sample was centrifuged (Hermle-Z233M-2 centrifuge, Labortechnik GmbH, Wasserburg, Germany) at 15,000 rpm for 30 min to sediment the cells, and the supernatant was extracted and used as a crude biodefoamer. All experiments were conducted in triplicate.

2.4. Response Surface Methodology

Response surface methodology (RSM) was used to determine optimum pH and concentration for the biodefoamer production, i.e., foam reduction efficiency and decay rate, for either the bio- or synthetic defoamers. Foam reduction efficiency and foam decay rate were the output variables. The design of experiments in RSM was used to generate various experimental conditions (pH 7–10) and concentrations (1–4% v defoamer/v MLSS and PSW mixture) [13,14]. A two-factor, five-level central composite design with 13 experimental runs was carried out. A T-test was conducted to evaluate the statistical importance of the regression coefficient to estimate biodefoamer production. A Fisher (F) test was employed to ensure the precision of the model obtained, i.e., to describe optimal conditions for biodefoamer production. The determination coefficient (R^2) was determined to assess the appropriateness of the model which described optimal biodefoamer production. See Table 1 for coded levels and descriptions.

Table 1. Media constituents included in the central composite design (CCD) experiments and their corresponding high, medium and low pH and concentration.

Variables Code Units High Level (+) Medium (0) Low Levels (–)

pH A - 10.60 8.50 6.30
Concentration B % (v/v) 4.60 2.50 1.00

2.5. Foaming Behavior Tests

2.5.1. Foam Reduction Efficiency

The PSW used in this study was collected from a slaughterhouse near Cape Town, South Africa, using 25 L polypropylene bottles, transported to the laboratory at 4 °C, and used when received. The quality of this wastewater varied considerably based on the environmental conditions; however, the characteristics of the wastewater used in the current study were comparable to other studies (see Table 2). A volume (100 mL) of PSW and mixed liquor suspended solids (MLSS) at a ratio of 1:2 was added to 250 mL graduated cylinders (foaming reactor, see Figure 1; the specifications of this reactor are tabulated in Table 3) after pneumatic mixing through air sparging using Mott element 6500 diffusers (Mott element 6500, Mott Corporation, New York, NY, USA) at 40 mL/min. After that, the foam was generated by sparging air using a Resun air pump (Resun Ac 9906, Shenzhen Xing Risheng Industrial Co. Ltd., Baolong, China) attached to an airflow meter (Key instrument FR 2000 series, Rhomberg Instruments, Cape Town, South Africa) at 80 mL/min until 82 mL of foam was generated. The foam volume (measured using a graduated cylinder)

cylinders (foaming reactor, see Figure 1; the specifications of this reactor are tabulated in Table 3) after pneumatic mixing through air sparging using Mott element 6500 diffusers (Mott element 6500, Mott Corporation, New York, NY, USA) at 40 mL/min. After that, the foam was generated by sparging air using a Resun air pump (Resun Ac 9906, Shenzhen Xing Risheng Industrial Co. Ltd., Baolong, China) attached to an airflow meter (Key instrument FR 2000 series, Rhomberg Instruments, Cape Town, South Africa) at 80 mL/min

s intervals). The foaming behavior was compared using foaming tendency (FT), the foam until 82 mL of foam was generated. The foam volume (measured using a graduated cylinder) was recorded against time using a stopwatch (foam volume generation was recorded at 40 s intervals). The foaming behavior was compared using foaming tendency (FT), the foam reduction efficiency (FRE) and foam collapse rate (FCR)—see Equations (1) and (2). This method was used without modification, as reported elsewhere [15].

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FT = (Foam generated immediately after aeration (mL)) ÷ (Rate at which air and (2). This method was used without modification, as reported elsewhere [15].

is sparged into the sample (mL/min)); (1)

$$FT = (\text{Foam generated immediately after aeration (mL)}) \div (\text{Rate at which air}$$

is sparged into the sample (mL/min)); (1)

$$FRE (\%) = (1 - FT \text{ of a sample with defoamer}) \div (FT \text{ sample without}$$

defoamer) × 100. (2)

$$FRE (\%) = (1 - FT \text{ of a sample with defoamer}) \div (FT \text{ sample without}$$

$$\text{defoamer}) \times 100. (2)$$

Table 2. Characteristics of poultry slaughterhouse wastewater from the literature reviewed and the current study.

Parameters	Minimum	Parameters	Minimum	Average	
		Maximum		(mg/L) References	Current
	(mg/L)		(mg/L)		Current Study (mg/L)
	(mg/L)		Average		Study (mg/L)
	Maximum		(mg/L) References		
				FOG 131 684 406 [16]	710–1178
FOG	131 684 406 [16]	710–1178	Proteins 0 368 184 [17]	29–96	
Proteins	0 368 184 [17]	29–96	TSS 198 3100 1207 [18]	2000–10,000	
TSS	198 3100 1207 [18]	2000–10,000	COD 2517 12,490 5216 ± 2534 [19]	4510–8355	
			COD 2517 12,490 5216 ± 2534 [19]	4510–8355	

Figure 1. Aeration column setup.

Figure 1. Aeration column setup.

Table 3. Foaming assessment specifications of the reactor.

Dimensions Specifications Cylinder

Material Polypropylene Height 24.5 cm Diameter 3 cm Air flow rate 80 mL/min

Diffusers

Porous length 2.32 cm Porous diameter 1.2 cm Media grade 40 Porous material 316LSS

For the foam collapse rate (FCR) (Equation (3)), a method by Mangundu [20] was used with minor modifications. A volume (100 mL) of PSW and mixed liquor-suspended solids (MLSS) at a ratio of 1:2 was used. This mixture was placed in a foaming reactor (250 mL graduated cylinder, see Figure 1 and Table 2 for dimensions). The height of the mixture was measured before the generation of foam. To thoroughly mix the PSW and MLSS, 40 mL/min of air was sparged through a porous air diffuser. The foam was generated by sparging air at a flow rate of 80 mL/min until a foam height of 13 cm was generated. The foam height collapse rate in the graduated cylinder was measured as a function of time using a stopwatch. The air pump (Resun Ac 9906, Shenzhen Xing Risheng Industrial Co. Ltd., Baolong, China) was switched off for 40 s such that the foam and liquid layer were distinguishable. A 4% (v/v) of synthetic defoamer, i.e., active silicone polymer antifoam A (A6582, Sigma-Aldrich, Unit 16/17 Lake Site, Industrial Park, Jet Park, SA), determined to be an efficient defoamer for WWTPs, was used at different pH and defoamer

concentrations that the RSM generated in combination with the PSW/MLSS mixture. After that, 40 mL/min of air was sparged through the samples again for mixing, and the foam collapse rate was measured every 10 s until the foam had decayed. If not, the foam height was regenerated in repeat tests. These tests were performed using triplicate samples, and an average of the samples was used to generate experimental data.

$$\text{FCR (mm/s)} = (\text{foam in the column after sparging}) \div (\text{time taken to collapse the entire foam}). \quad (3)$$

2.5.3. Microscopic Analysis of Sludge Agglomeration in the Presence and Absence of Bio and Synthetic Defoamers

AS samples were collected from a WWTP (Cape Town, South Africa) using 4 L polypropylene bottles. The sludge samples were taken from the aeration tank and transported to the laboratory. To ensure aerobic conditions, the sampling polypropylene bottles were partly filled (one-third) and tightly closed after refrigeration at 4 °C. Before experimental tests, the bottles were swirled to mix the sludge, and the agglomerated biomass was

analyzed according to Eikelboom et al. [21]. The filamentous microorganisms observed were compared with the images depicted in the method. A filamentous index (FI) (the size of filamentous bacteria population observed using a microscope, 0–5 non-numerous filaments) was quantified. An experiment was conducted to observe the effect of the biodefoamer and synthetic defoamer on the AS filamentous bacteria. To observe this effect, a 96-well tissue culture plate (Corning® 96-well clear polystyrene microplates, Merck, Rahway, NJ, USA) was used, with four wells serving as a control. The experiment was conducted to mix 200 µL of the MLSS with 160 µL of sterile PSW filtered through a 0.22 µm membrane filter and added to the four wells. After that, a volume (245.6 µL) of the MLSS was added to each well, after which 100 µL of the sterile PSW was added in the subsequent wells ($n = 8$). Out of the eight wells, four wells were supplemented with 14.4 µL of the biodefoamer, the other four were supplemented with a volume of 14.4 µL of the synthetic defoamer was used for comparative analysis. This was followed by incubation at 37 °C for ten days. Afterward, 25 µL of each sample was extracted from each well on days 0, 5 and 10. The samples were gram stained and viewed under a light microscope (Olympus CX21FS1 microscope, Olympus Corporation, Tokyo, Japan) at 1000×. The microscope was connected to a computer running Dino Capture 2.0 software (Dino-Lite digital microscope, London, England, UK), using a 4000 µm and 10 µm scale bar. This was to observe the effect of each defoamer on the MLSS microbial population.

2.6. Characterization of the Biodefoamers

The consortia were further cultured in the aforementioned biodefoamer production media and incubated for 48 h at 120 rpm at an optimum pH and temperature determined using RSM. After that, the consortia were put on ice and centrifuged at 10,000 rpm for

30 min, with the supernatant being collected for processing using ethanol (4 °C) at a ratio of 2:1 (v/v), that is, alcohol:supernatant, followed by shaking the mixture at 121 rpm at 4 °C overnight to produce a precipitate. The precipitate was then collected and re-centrifuged using a benchtop centrifuge at 15,000 rpm for 30 min. This procedure was repeated

($n = 3$ times) to collect a substantial quantity of the precipitate for each sample, after which the precipitate was washed using sterile distilled water and dialyzed against the water overnight, subsequent to vacuum drying in a desiccator for 24 h. The dried precipitate was analyzed using a Fourier transform infrared spectrophotometer (FTIR) (Spectrum two, PerkinElmer, Waltham, MA, USA).

Additionally, nuclear magnetic resonance spectroscopy analysis was conducted using the purified and dried precipitate by dissolving it in 1 mL of deuterium oxide (D_2O). This was followed by swirling for 5 min and then left overnight at ambient temperature. The swirled solution was then transferred into a 5 mL tube which was then capped and analyzed using 1H NMR (Bruker 400, Bruker Nano GmbH, Berlin, Germany) at a frequency resonance of 400 MHz.

3. Results and Discussion

3.1. Microbial Isolation and Identification of Biodefoamer-Producing Isolates

Amongst the microorganisms that were isolated from the PSW discharge spout, four isolates, i.e., *Bacillus subtilis* (GCA_000009045.1), *Aeromonas veronii* (GCA_000204115.1), *Klebsiella grimontii* (UGJQ01000001.1) and *Comamonas testosteroni* (GCA_900461225.1) were selected for biodefoamer production based on their rapid efficiency in foam decay, moderate pH and lower concentration. The *Bacillus* sp. colonies were rod-shaped and off-white, and when they were viewed under a microscope, their gram staining depicted that they were gram-positive. *Aeromonas veronii* was also off-white, albeit gram-negative. *Klebsiella grimontii*'s gram reaction was negative, with rod-shaped, slimy and white colonies. Similarly, *Comamonas testosteroni*'s gram reaction was negative, exhibiting slightly curved rods with

mucoïd colonies, which were also white. The consortium produced biodefoamers under various environmental conditions as determined using response surface methodology. This was the first study to use a consortium isolated from poultry slaughterhouse wastewater discharge spout to produce biodefoamers.

3.2. Mixed Liquor Suspended Solids (MLSS) Metagenomics Analysis

The AS microorganism sequences were generated using metagenomics, indicating that the dominant foam formers in the AS samples were *Nostocoida limicola*, *Gordonia krop penstedtii*, *Candidatus Microthrix parvicella*, *Nocardioides insulae* and *Bacteroides nordii*. These

biofoamers are hydrophobic actinobacteria that contain mycolic acids and produce biosurfactants to break down FOG so that they uptake it as a carbon source. These characteristics

enhance biofoamation in AS systems [22,23].

3.3. Biodefoamer Production, Reactor Conditions Optimization and Characterization

Two independent variables, pH (range 7–14) and concentration (range 1–4% v/v), were used to evaluate the defoamer (synthetic and biodefoamer) activity (foam reduction efficiency and foam decay rate). The efficiency of biodefoamers produced from a competitive consortium was compared to that of the synthetic defoamer. The results in Table 4 showed that pH 7 at a concentration of 4% (v/v) positively affected biodefoamer production, which inculcated rapid foam decay and suppression rate. The results depicted that the biodefoamer effect became less effective as the pH increased. This was because the microorganisms preferred a neutral pH to produce the biodefoamers with high activity. Overall, biodefoamers have a different electrical state at different pHs. A change in pH not only alters the defoamer's charge, it also alters the charge of the suspended solids, which affects the settleability of the MLSS and the PSW SS, which in turn affects the foaming behavior of the wastewater [24]. The concentration of the biodefoamer is also crucial. Low concentration affects the bridging mechanism. Higher concentration leads to sludge

deflocculation and high viscosity, which further results in foam formation; hence, lower and higher concentrations produce minimal biodefoamer activity [25]. According to the synthetic defoamer specification, it is effective at a concentration of 1–100 ppm and stable at pH 5–9. According to this study, it was more effective at pH 7 and 1% (v/v), and this falls in the specific ranges of the defoamer specification. This defoamer's effectiveness also deteriorated at higher pH, albeit at a lower concentration, because silicone-based defoamers are sensitive to pH. They lose their activity at acidic pH and pH more significant than 10 [26]. These results highlight that the defoamers are not universal because their

activity depends on the ever-changing environmental condition of the CAS. Table 2 lists the ANOVA quadratic model used to optimize biodefoamer production.

Table 4. Experimental design table for the independent variables pH (A) and defoamer concentration (B); and for the dependent variables (foam reduction efficiency and foam collapse rate of the biodefoamer and synthetic defoamer (AB)).

Run	pH	Concentration (%v/v)	Foam Reduction Efficiency		Foam Collapse Rate (mm/s)	
			Defoamer	Biodefoamer	Synthetic	Synthetic
1	8.5	0.38	93	73	0.83	0.5
2	10	1	2.9	73	0.55	0.55
3	8.5	2.5	83	82	0.42	0.5
4	8.5	2.5	83	82	0.42	0.5
5	8.5	4.6	15	15	0.5	0.5
6	3.8	2.5	96	42	1	0.5
7	8.5	2.5	83	82	0.42	0.5
8	8.5	2.5	83	82	0.42	0.5
9	10.6	2.5	87	73	0.45	0.45
10	8.5	2.5	83	82	0.42	0.5
11	7	1	90	73	1.25	0.83
12	7	4	96	96	1.7	2.5
13	10	4	95	78	0.42	0.42

The RSM was used to optimize the environmental conditions of the bioreactor such that an optimum response was achieved. The effect of the two independent variables (pH and concentration) on the foam reduction efficiency as well as foam decay rate, which were used to determine biodefoamer production, are graphically illustrated in Figure 2A for the biodefoamer and in Figure 2B for the synthetic defoamer. It was observed that both pH and concentration significantly impacted the efficacy of both defoamers. Wongsamuth and Doran [27] also reported that pH and defoamer concentration significantly impact a defoamer's effectiveness. The optimal biodefoamer production was observed at pH 7 and a concentration of 4% (v/v), where the highest foam reduction efficiency was 96% at a faster foam decay rate of 1.7 mm/s. The biodefoamer results are similar to the results Mamais [28] obtained when a polyaluminum chloride (PAX-14) concentration of 6.6 to 11.5 g Al³⁺/kg MLSS reduced foam by 75 to 100%, improving the sludge flocculation. However, PAX could only eliminate *Gordonia amarae* and *Microthrix parvicella* cells, and using chlorine-based chemicals can be detrimental to the microbial communities because it can break down the microbial cell walls.

The lowest biodefoamer production was observed at pH 8.5 and at a higher defoamer concentration of 4.6 (v/v), with a foam reduction efficiency of 15% at a decay rate of 0.42 mm/s. This showed that the defoamer production was more affected by the pH than the defoamer concentration. Similarly, the biodefoamer obtained a minute foam collapse rate of 0.42 mm/s at alkaline conditions pH 10, even with a higher defoamer concentration of 4% (v/v); however, the foam reduction efficiency of 78% was observed. This was due to the deprotonation of the biodefoamer amino charges under alkaline conditions.

Figure 2. A graphical illustration showing the effect of pH and concentration on (A,C) foam reduction efficiency of the biodefoamer; synthetic defoamer (B,D) foam collapse rate in the presence of a biodefoamer and a synthetic defoamer, respectively.

The highest foam reduction efficiency for the synthetic defoamer was 96%, which corresponded with the highest foam collapse rate of 2.5 mm/s at a pH of 7 and a higher

The foam generation rate is usually faster than the foam collapse rate [28]. The foam concentration of 4% (v/v). The lowest foam reduction efficiency of the synthetic defoamer height reduction over time was recorded in this study. The color mash red (in this case was 2.9% at pH ten and a concentration of 1% (v/v). In contrast, the lowest foam decay dark orange) is the highest optimal color and the blue is the lowest. Figure 3 mash color rate of 0.42 mm/s was observed at pH 10 and a concentration of 4% (v/v). This indicated is dark orange to red When pH is 7 and the concentration is 4 %v/v it indicates maximum that the synthetic defoamer was ineffective at very alkaline pH, regardless of the defoamer foam reduction efficiency. It becomes blue at increased pH of up to 10. Figure 3 shows concentration [30]. Overall, the synthetic defoamer had a higher foam collapse rate than that the foam collapse rate is minimal at elevated pH, concentration as well as pH below the biodefoamer, although foam reduction efficiency was similar.

7. Figure 3 illustrates the foam behavior in the absence and presence of both a bio- and synthetic defoamer (4% v/v) at pH 7. For both the bio- and synthetic defoamer, the foam production. The mean was 0.4, and the standard deviation for the model was 0.3. Essential decay rate was observed to be under 40 s. After that, there was minimal foam regeneration parameters in the model had $p < 0.05$; however, the parameters with a p value that was even with subsequent sparging, meaning that both defoamers were efficient in foam re greater than the latter were negligible, thus indicating that the terms A and A^2 were of duction. However, the biodefoamer reduced the foam height by 85.4% within 40 s, significance; hence, Equation (4) was simplified to Equation (5).

whereas the synthetic defoamer only reduced the foam height by 61.5%. Mangundu [20] assumed that the minimum threshold required for foam decay was above 30%; however,

$$Y = 0.42 + 0.34A - 0.02B - 0.15AB - 0.22A^2 - 0.19B^2; \quad (4)$$

in this study, a defoamer concentration of 4% (v/v) was able to collapse and suppress the foam height significantly. The efficiency of a defoamer is determined by its potential to

$$Y = 0.42 + 0.22A^2. \quad (5)$$

form a destabilizing bridge that stretches across the lamellae such that it ruptures individual foam bubbles, thus causing liquid drainage from the foam, causing the foam to col

Equation (6) was used to estimate the effectiveness of synthetic defoamers and foam lapse. Moreover, it was previously demonstrated that bubble coalescence occurs in such collapse rate. This model was also deemed to be insignificant to adequately describe the instances; therefore, both defoamers had a potential to be used in large-scale applications foam reduction efficiency, with a mean of 71.7 as well as a standard deviation of 23.1, and [31]. Overall, when no defoamers were applied, the foam height was slightly reduced for an F-value that was more than 0.57; the *p* values for all the terms in the model were more than 0.05. The foam height was reduced by 23%. The constituents of the wastewater, such as

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significant than 0.05, showing this quadratic model to be insignificant in predicting the foam reduction efficacy of synthetic defoamers.

$$Y = 155 + 44A + 38B - 02AB - 2.2A^2 - 5B^2. \quad (6)$$

3.4. Dynamic Foam Decay Test

The foam generation rate is usually faster than the foam collapse rate [28]. The foam height reduction over time was recorded in this study. The color mash red (in this case dark orange) is the highest optimal color and the blue is the lowest. Figure 3 mash color is dark orange to red When pH is 7 and the concentration is 4 %v/v it indicates maximum foam reduction efficiency. It becomes blue at increased pH of up to 10. Figure 3 shows that the foam collapse rate is minimal at elevated pH, concentration as well as pH below 7. Figure 3 illustrates the foam behavior in the absence and presence of both a bio- and synthetic defoamer (4% v/v) at pH 7. For both the bio- and synthetic defoamer, the foam decay rate was observed to be under 40 s. After that, there was minimal foam regeneration even with subsequent sparging, meaning that both defoamers were efficient in foam reduction. However, the biodefoamer reduced the foam height by 85.4% within 40 s, whereas the synthetic defoamer only reduced the foam height by 61.5%. Mangundu [20] assumed that the minimum threshold required for foam decay was above 30%; however, in this study, a defoamer concentration of 4% (v/v) was able to collapse and suppress the foam height significantly. The efficiency of a defoamer is determined by its potential to form a destabilizing bridge that stretches across the lamellae such that it ruptures individual foam bubbles, thus causing liquid drainage from the foam, causing the foam to collapse.

Moreover, it was previously demonstrated that bubble coalescence occurs in such instances; therefore, both defoamers had a potential to be used in large-scale applications [31]. Overall, when no defoamers were applied, the foam height was slightly reduced for 210 s. The

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foam height was reduced by 23%. The constituents of the wastewater, such as dissolved solids, including the presence of microorganisms such as *Actinomyces* sp. in the MLSS, which produce biosurfactants and reduce the surface tension, can increase the viscosity of the MLSS, which produces biosurfactants and reduce the surface tension, can increase the viscosity that stabilizes the lamellae, increasing foam surface elasticity. This can further increase bubble deflocculation, causing excessive foaming as minimal pressure will be applied on the gas–liquid interface, resulting in minute liquid drainage [32]. When such a thing occurs in a WWTP, excessive foam generation will ensue, causing a nuisance for operators and facility managers.

Figure 4. A micrograph representation of sludge: in the absence of a defoamer (A–C), in the presence of a biodefoamer (D–F) and when it was exposed to a synthetic defoamer (images G–I).

3.6. Biodefoamer FTIR and ^1H NMR Characterization

The synthetic defoamer, however, had a negative effect on sludge compaction. The Fourier infrared spectroscopy ((FTIR) of the biodefoamer that was produced images showed oil droplets assumed to be from FOG, which significantly contributes to a PSW consortium in Figure 5 was conducted to examine the correlation of the functional filamentous foam-forming bacterial growth. Observations indicated that on day 0, there were groups to biodefoamer activity. The peak at 3292 (49.20%; 3292 cm^{-1}) indicated an excessive filamentous bacterial growth. On day 5, although there was minute floc functional group [38,39], while the peak at 2112.25 (95.62%; 2112.25 cm^{-1}) [38], [39] indicated formation and visible oil droplets, the filaments protruded outside of individual flocs, an aliphatic C-H stretching group [39]; the peak at 1634. Seventy-five (70.27%; 1634 leading to bridging and recalcitrant deflocculation on day 10, as observed elsewhere [34]. indicated N-H- bending primary amines or carboxylic groups [40,41]. These groups. These results showed that the number of filamentous bacteria must not be excessive, as this carboxyl, alkane, amine and hydroxyl groups, confirmed that the biodefoamer will lead to weaker floc agglomeration. The synthetic defoamer favored the proliferation of dominantly a polysaccharide. Although it was concluded that the biodefoamer was some, if not most, filamentous bacteria associated with WWTP [35]. It was demonstrated sensitive, minute concentrations were required for its efficacy, imparting electrical that the non-application of a defoamer in the control experiments resulted in the MLSS neutrality to MLSS; however, the crude biodefoamer contained impurities, all of which showing a significant proliferation of filamentous bacteria throughout the experiment until sulted in foam decay.

day 10, with minute floc formation or agglomeration being observed. This was attributed The weak peaks at 410–439 (30.48–31.19%; 410–46–439.18 cm^{-1}) were associated to the changes in the quality characteristics of wastewater whereby aged sludge (>27 days) azomethine groups $\text{CH}=\text{N}/\text{C}=\text{N}$ [42]. These groups, i.e., carboxyl, alkane, amine and can be easily disintegrated with a weakened compaction attribute due to pH, organic hydroxyl groups, confirmed that the biodefoamer was predominantly a polysaccharide and inorganic nutrients present therein [12]. Previously, it was observed that excessive carboxyl functional groups contain multiple binding sites, ensuring compact attachment filamentous growth leads to poor sludge sedimentation, the washout of microbial cells through charge neutralization and/or bridging [11]. In charge neutralization, the from the aeration tank/stage in the WWTP, and excessive biofoam formation [36,37]. tively charge MLSS particulates will bind to the opposite charge in the defoamer, and will reduce the repulsive electrostatic forces and increase attraction forces, meaning the defoamer will adsorb and neutralize the charge of the MLSS particulate. Although was concluded that the biodefoamer was pH-sensitive, minute concentrations were required for its efficacy, imparting MLSS foam electrical neutrality; however, the crude defoamer contained impurities, all of which resulted in foam decay.

3.6. Biodefoamer FTIR and ^1H NMR Characterization

The Fourier infrared spectroscopy ((FTIR) of the biodefoamer that was produced by a

PSW consortium in Figure 5 was conducted to examine the correlation of the functional groups to biodefoamer activity. The peak at 3292 (49.20%; 3292 cm^{-1}) indicated an OH- functional group [38,39], while the peak at 2112.25 (95.62%; 2112.25 cm^{-1}) [38], [39] indicated an aliphatic C-H stretching group [39]; the peak at 1634.75 (70.27%; 1634 cm^{-1}) indicated N-H- bending primary amines or carboxylic groups [40,41]. These groups, i.e., carboxyl, alkane, amine and hydroxyl groups, confirmed that the biodefoamer was predominantly a polysaccharide. Although it was concluded that the biodefoamer was pH-sensitive, minute concentrations were required for its efficacy, imparting electrical neutrality to MLSS; however, the crude biodefoamer contained impurities, all of which resulted in foam decay.

Figure 5. Fourier transform infrared spectroscopy (FTIR) spectrogram of the biodefoamers produced by a PSW consortium.

The weak peaks at 410–439 (30.48–31.19%; $410.46\text{--}439.18\text{ cm}^{-1}$) were associated with

The ^1H NMR spectrum in Figure 6 revealed signals between δ 0.8–1.9 ppm, confirming azomethine groups $\text{CH}=\text{N}/\text{C}=\text{N}$ [42]. These groups, i.e., carboxyl, alkane, amine and hydroxyl groups, confirmed that the biodefoamer was predominantly a polysaccharide. The chemical shifts from δ 2.0–2.92 ppm showed the presence of amines, whereas the signal at carboxyl functional groups contain multiple binding sites, ensuring compact attachment

δ 2.0 ppm showed CH- protons that were bounded to a vinyl carbon the solvent peak through charge neutralization and/or bridging [11]. In charge neutralization, the negatively appeared at δ 4.7 ppm, and the signal at δ 3.1 ppm showed the presence of a hydroxyl charge MLSS particulates will bind to the opposite charge in the defoamer, and this will group [13]. The concentrated conspicuous peaks revealed the presence of aromatic rings reduce the repulsive electrostatic forces and increase attraction forces, meaning that the in the structure. The multiple signals between δ 3.5 and 4.0 ppm were carbohydrate at defoamer will adsorb and neutralize the charge of the MLSS particulate. Although it was tributes, whereas the signals from δ 0.8 to 3.5 ppm were similar to a polysaccharide structure. The chemical shifts between δ 3.5 to δ 4.0 revealed that the polysaccharide structure for its efficacy, imparting MLSS foam electrical neutrality; however, the crude biodefoamer had an electro-negative oxygen that originated from an ester functional group attached to contained impurities, all of which resulted in foam decay.

it, resulting in functional groups such as alkoxy and hydroxyl [24]. The carbohydrate sig

The ^1H NMR spectrum in Figure 6 revealed signals between δ 0.8–1.9 ppm, confirming signals usually resonated between δ 3.5 and 4.5 ppm; some carbohydrate peaks were observed the aliphatic stretches' presence and the signal at 1.9 ppm depicted a CH_2 proton. The served from δ 4.4 to 5.5 ppm, and others appeared from δ 0 to 5 ppm. These are unique chemical shifts from δ 2.0–2.92 ppm showed the presence of amines, whereas the signal from the existing literature, and these peaks are affected by various factors, such as the

at δ 2.0 ppm showed CH- protons that were bounded to a vinyl carbon the solvent peak type of solvent used to dissolve the sample [29]. The ^1H NMR spectra can be used to estimate the configuration of the glycosidic bonds of the polysaccharide structure. When the signal at δ 4.7 ppm, and the signal at δ 3.1 ppm showed the presence of a hydroxyl group [13]. The concentrated conspicuous peaks revealed the presence of aromatic rings in

chemical signals of a proton are more significant than 5 ppm, the structure is predominantly α glycosidic linkage. However, if it is less than 5 ppm, the glycosidic bonds are mostly β . The biodefoamer used in this study is a carbohydrate with predominantly β glycosidic linkages. The carbohydrate structure of the biodefoamer will enhance MLSS flocculation due to its surface charges.

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The chemical shifts between δ 3.5 to δ 4.0 revealed that the polysaccharide structure had an electro-negative oxygen that originated from an ester functional group attached to it, resulting in functional groups such as alkoxy and hydroxyl [24]. The carbohydrate signals usually resonated between δ 3.5 and 4.5 ppm; some carbohydrate peaks were observed from δ 4.4 to 5.5 ppm, and others appeared from δ 0 to 5 ppm. These are unique from the existing literature, and these peaks are affected by various factors, such as the type of solvent used to dissolve the sample [29]. The ^1H NMR spectra can be used to estimate the configuration of the glycosidic bonds of the polysaccharide structure. When the chemical signals of a proton are more significant than 5 ppm, the structure is predominantly α glycosidic linkage. However, if it is less than 5 ppm, the glycosidic bonds are mostly β . The biodefoamer used in this study is a carbohydrate with predominantly β glycosidic linkages. The carbohydrate structure of the biodefoamer will enhance MLSS flocculation due to its surface charges.

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14 of 17 The carbohydrate structure of the biodefoamer will enhance MLSS flocculation due to its surface charges.

Figure 6. ^1H NMR spectra of a biodefoamer.
Figure 6. ^1H NMR spectra of a biodefoamer.

4. Conclusions

4. Conclusions

The isolated microbial consortium of *Bacillus*, *Aeromonas*, *Klebsiella* and *Comamonas* spp.

The isolated microbial consortium of *Bacillus*, *Aeromonas*, *Klebsiella* and *Comamonas* spp. produced biodefoamers with high biodefoamer efficacy and a foam decay rate suitable for WWTP application when RSM optimized the application conditions. The highest synthetic and biodefoamer efficiency was 96% under these conditions, i.e., pH 7 and a

concentration
 synthetic and biodefoamer efficiency was 96% under these conditions, i.e., pH 7 and a of 4% (v/v). When further foam collapse studies were carried out, i.e., using the RSM concentration of 4% (v/v). When further foam collapse studies were carried out, i.e., using optimum conditions, the maximum foam decay rate for the synthetic and biodefoamer was

the RSM optimum conditions, the maximum foam decay rate for the synthetic and bio determined to be 99% at 50 min, with total foam suppression at 190 min. The microscopic defoamer was determined to be 99% at 50 min, with total foam suppression at 190 min. pictures indicated that the AS was impacted by the presence and absence of defoamers

The microscopic pictures indicated that the AS was impacted by the presence and absence (synthetic and biodefoamers), with the absence of both defoamers resulting in excessive

of defoamers (synthetic and biodefoamers), with the absence of both defoamers resulting filamentous bacterial growth. The FTIR spectra indicated that the biodefoamer used in this

in excessive filamentous bacterial growth. The FTIR spectra indicated that the bio study was a polysaccharide, and ¹H NMR confirmed that the defoamer was a carbohydrate defoamer used in this study was a polysaccharide, and ¹H NMR confirmed that the constituent. These characteristics of the biodefoamers produced by *Bacillus*, *Aeromonas*,

defoamer was a carbohydrate constituent. These characteristics of the biodefoamers produced by *Bacillus*, *Aeromonas*, *Klebsiella* and *Commamonas* spp. consortia show that it is competent for activated sludge that contains biofoamers. The results that are portrayed in this paper show that the consortium competent for activated sludge that contains biofoamers. The results that are portrayed in was isolated from poultry slaughterhouse wastewater and it produced biodefoamers that this paper show that the consortium was isolated from poultry slaughterhouse wastewater and it produced biodefoamers that were used for PSW and activated sludge defoamation. These biodefoamers' foam reduction efficiency and foam decay rate were comparable to that of a synthetic defoamer, and it did not just reduce foam but also flocculated the sludge to prevent continuous foam regeneration and dosing. The microscopic

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were used for PSW and activated sludge defoamation. These biodefoamers' foam reduction efficiency and foam decay rate were comparable to that of a synthetic defoamer, and it did not just reduce foam but also flocculated the sludge to prevent continuous foam regeneration and dosing. The microscopic analysis that was carried out in this study revealed that the synthetic silicone defoamer left oil residues in the sludge that harnesses filamentous growth and foam regeneration, which will require continuous dosing. This was the first study to compare a biodefoamer that is produced by a microbial consortium with a synthetic defoamer.

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